

Annulation Reaction of 3-Acylmethylidene Oxindoles with Huisgen Zwitterions and Its Applications in the Syntheses of Pyrrolo[4,3,2-de]quinolinones and Marine Alkaloids Ammosamides

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Supporting Information

ABSTRACT: A novel annulation reaction of 3-acylmethylidene oxindoles with Huisgen zwitterions is unveiled that leads to an unprecedented synthetic method for complex pyrrolo-[4,3,2-de]quinolinones and marine alkaloids ammosamides A—C. This method features simplicity, high efficiency, and broad substrate scope and is accordingly anticipated to significantly facilitate the preparation and bioassay of the relevant pyrroloquinoline alkaloids and their analogues.

The pyrrolo [4,3,2-de] quinoline core represents a characteristic motif of many natural alkaloids that exhibit a wide range of biological activities, including potent cytotoxicity and pharmacological activity. Shown in Figure 1 are some typical

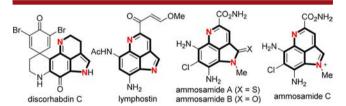


Figure 1. Representative natural pyrroloquinoline alkaloids.

examples of pyrroloquinoline alkaloids with attractive bioactivities. 2-5 Isolated early from marine sponges, discorhabdin C possesses potent cytotoxicities against several cancer cell lines such as murine leukemia cells P388 and L1210 and human colon cancer cells HCT-116.2 Lymphostin as a natural immunosuppressant was first isolated from bacteria Streptomyces sp. KY11783 and exhibits potent inhibitory activities against lymphocyte kinase (IC₅₀ 0.05 μ M) and phosphatidylinositol 3kinase (IC₅₀ 0.001 μ M). Recently isolated from marine bacteria Streptomyces strain CNR-698, ammosamides A-C as new members of pyrroloquinoline alkaloids are able to influence tubulin and actin dynamics through myosin targeting. Ammosamides A and B exhibit significant cytotoxicity against HCT-116.46 Ammosamide B and its methylated analogue have potent inhibitory activity against human quinone reductase II.5 Owing to their promising biological properties and novel molecular structures, the pyrroloquinoline alkaloids have attracted a great deal of attention from the chemistry community in recent decades. Many effective multistep synthetic strategies have already been developed in the total syntheses of some typical natural alkaloids. 16,e,f,6 However, efficient and flexible

synthesis of this family of alkaloids and their analogues remains a challenging goal. A lack of simple and amendable synthetic methodology still hinders pyrroloquinoline alkaloids from the optimization of their biological activity and other relevant investigations. The development of general and efficient synthetic methods is therefore highly desirable.

The Huisgen zwitterions, readily generated from Ph₃P and dialkyl azodicarboxylates, are well known as the key intermediate in the classical Mitsunobu reaction and early reported annulation reactions. Recent renewed interest from Nair group and others has unveiled the attractive versatility of Huisgen zwitterions in the C–N bond-forming reactions. In particular, in an array of annulation reactions with various electrophiles like carbonyls, imines, and electron-deficient alkenes, Huisgen zwitterions effect unique and efficient routes to aza-heterocycles (Scheme 1). 11,12

Scheme 1. Annulation Modes of Huisgen Zwitterions

$$\begin{array}{c} \text{PPh}_3 \\ + \text{CO}_2 \text{R} \\ \text{N=N} \end{array} \xrightarrow{\begin{array}{c} \text{Ph}_3 \text{P} \\ \text{RO}_2 \text{C} \end{array}} \begin{array}{c} \text{CO}_2 \text{R} \\ \text{N-N} \\ \text{RO}_2 \text{C} \end{array} \xrightarrow{\begin{array}{c} \text{Ph}_3 \text{PO} \\ \text{Ph}_3 \text{PO} \end{array}} \begin{array}{c} \text{X = O, NR', CR'R''} \\ \text{Ph}_3 \text{PO} \\ \text{OR} \end{array}$$

Intrigued by the unique annulation modes of Huisgen zwitterions and potential biological activity of oxindole derivatives, we very recently explored the possible annulation reaction between 3-acylmethylideneoxindoles 1 and in situ generated Huisgen zwitterions from azodicarboxylates 2 and Ph₃P (Table 1). To our surprise, an unexpected annulation reaction occurred, leading to dihydropyrrolo[4,3,2-de]-quinolinones 3. This reaction does unveil a novel annulation

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Table 1. Annulation Reaction between 3-Acylmethylidene Oxindoles 1 and Huisgen Zwitterions^a

entry	X, R ¹ , R ² in 1	R in 2	3 ^b (%)	4 ^b (%)
1	H, Bn, Ph (1a)	Et (2a)	3a, 84	4a , 13
2	$H, Bn, 4-FC_6H_4(1b)$	2a	3b , 63	4b , 30
3	H, Bn, CO ₂ Et (1c)	2a	3c, 82	
4	1c	2a	3c, 99 ^c	
5	H, Me, CO_2Et (1d)	2a	3d , 89 [€]	
6	H, allyl, CO ₂ Et (1e)	2a	3e, 99 ^c	
7	H, Boc, CO ₂ Et (1f)	2a	3f, 82°	
8	6-Br, Bn, CO ₂ Et (1g)	2a	3g, 85°	
9	H, Bn, Me (1h)	2a	complex	
10	la	<i>i</i> -Pr (2b)	3h , 83	4c, 11
11	la	<i>t</i> -Bu (2c)	3i, 96	trace
12	la	Bn (2d)	3j , 94	trace
13	1c	2b	3k, 94°	
14	1c	2c	3 l , 90°	
15	1c	2d	3m, 95 ^c	
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^aTypical conditions: under a N_2 atmosphere, to a stirred mixture of 1 (0.5 mmol) and 2 (0.75 mmol) in CH_2Cl_2 (5.0 mL) was added Ph_3P (0.75 mmol), and the resulting mixture was stirred at rt for 15 min to 10 h until 1 was completely consumed. For details, see the Supporting Information. ^bIsolated yields. ^cTHF was used instead of CH_2Cl_2 .

mode of Huisgen zwitterions involving a rare N–N bond cleavage and also provides an unprecedented one-step strategy to construct a complex pyrrolo[4,3,2-de]quinoline skeleton. Further investigations have resulted in a simple and highly efficient synthetic methodology for a variety of pyrrolo[4,3,2-de]quinolinones and natural alkaloid ammosamides A–C as well. Herein, we communicate the relevant results.

Initially, we surveyed the model reaction between 3-(benzoylmethylidene) oxindole ${\bf 1a}$, diethyl azodicarboxylate ${\bf 2a}$, and ${\rm Ph_3P}$ for optimal conditions by varying the ratio of reactants and the solvent (for details, see the Supporting Information). Under a nitrogen atmosphere, the reaction of ${\bf 1a}$ (0.5 mmol), ${\bf 2a}$ (0.75 mmol), and ${\rm Ph_3P}$ (0.75 mmol) in a solvent (5.0 mL) like ${\rm CH_2Cl_2}$, ${\rm CHCl_3}$, toluene, acetonitrile, 1,4-dioxane, and THF readily delivered the annulation product ${\bf 3a}$ in fair to good yields after being stirred at rt for 15 min to 24 h. A concurrent minor product ${\bf 4a}$ was also collected in less than 25% yields. The best yield (84%) of ${\bf 3a}$ was obtained from the model reaction run in ${\rm CH_2Cl_2}$ at rt for 15 min with the minor product ${\bf 4a}$ in 13% yield (Table 1, entry 1).

The scope of the reaction was investigated under the preferred conditions (Table 1). With azodicarboxylate 2a (R = Et) used as a representative reactant, a selection of 3-acylmethylidene oxindoles 1 were tested. The aryl-substituted oxindoles 1 like 1b ($R^2 = p ext{-}FC_6H_4$) readily gave the annulation product 3b in 63% yield and minor product 4b in 30% yield (entry 2). Functional group ester-substituted oxindoles 1 ($R^2 = CO_2Et$) proved to be effective in the annulation reaction, giving the products 3 exclusively (Table 1, entries 3-8). Under the typical conditions, the reaction of 1c and 2a readily afforded 3c as a sole product in 82% yield (entry 3). Running the reaction in THF even brought about a 99% yield (entry 4). Other typical ester-substituted oxindoles 1d-g all exclusively delivered their

annulation products 3 in good to excellent yields (entries 5–8). However, methyl-substituted oxindole 1h ($R^2 = Me$) proved to be ineffective in the reaction with 2a (entry 9). Choosing 1a as one reactant, different azodicarboxylates 2 were also examined (entries 10-12). Diisopropyl azodicarboxylate (2b) readily gave the annulation product 3h in 83% yield and minor product 4c in 11% yield (entry 10). In contrast, azodicarboxylates 2c (R = t-Bu) and 2d (R = Bn) both delivered their annulation products 3 in excellent yields as the sole separable products (entries 11 and 12). In the annulation reactions with ester-substituted 1c, azodicarboxylates 2b-d, just like 2a, all delivered their exclusive products 3 in high yields (entries 13-15).

As shown in Table 1, aryl-substituted oxindoles 1 are effective substrates, affording dihydropyrroloquinolinones 3 as the major product (Table 1, entries 1 and 2). However, an interesting exception was observed: under similar conditions, o-tolyl-substituted oxindole 1i ($R^2 = o$ -tolyl) failed to deliver the expected product but a spirooxindole—pyrazoline 5 in modest yields (eq 1). Presumably, the considerably increased steric effect

of *o*-tolyl group hindered its adjacent carbonyl from involvement in the expected annulation reaction, and consequently, the reaction was led to a distinct but known pathway.¹³

To discern clues about the formation of product 4, we treated 3a in THF with a catalytic amount of amine base 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) at rt for 1 h and found that 3a was completely converted to 4a in 99% yield (eq 2). This result shows the annulation product 3 is convertible to the concomitant product 4 under base catalysis.

Although a precise mechanism of this annulation reaction remains elusive, on the basis of our results and closely related reports, ^{14–16} a proposed reaction is exemplified in Scheme 2. The reaction sequence should be initiated with the in situ generation of Huisgen zwitterion. ¹⁷ As a nucleophile, Huisgen zwitterion engages in a nucleophilic addition to the ketone carbonyl of 3-acylmethylidene oxindole **1a** to form intermediate **A**, which then eliminates triphenylphosphine oxide to produce

Scheme 2. Proposed Mechanism of the Annulation

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diaziridine B. ¹⁴ Under the intramolecular attack of the electronrich benzene ring, highly strained diaziridine B undergoes a heterolytic fission of the N–N bond, generating tricyclic intermediate C. ^{15,16} Through a proton transfer, C readily furnishes the annulation product 3a via a possible intermediate D. In another scenario, intermediate D undergoes an intramolecular oxo-Michael addition, followed by a proton transfer and a ring-opening elimination to generate the concomitant product 4a.

To expand its synthetic application of the annulation reaction to the challenging pyrrolo [4,3,2-de] quinolinone core, we further investigated the aromatization transformation of product 3. Considering the susceptibility of Boc group to the acid-triggered decomposition, we chose the annulation product 3i as a substrate and treated it with the catalytic amount of aq $\rm H_2SO_4$ (2.5 M, 20 mol %) in methanol at rt for 30 min (Scheme 3). To our delight, the desired aromatization product 6a was obtained in almost theoretical yield.

Scheme 3. Aromatization Transformation of 3i into 6a

With this exciting result in hand, we devised a two-step, one-pot synthesis of pyrrolo[4,3,2-de] quinolinones 6 from 3-acylmethylidene oxindoles 1 and azodicarboxylate 2c (Scheme 4). Under the typical conditions for the annulation reaction, 1 and 2c reacted at rt until 1 was completely consumed. The reaction mixture was then treated with the catalytic amount of H_2SO_4 , yielding the desired product 6 (for details, see the

Scheme 4. One-Pot Synthesis of Pyrrolo [4,3,2-de] quinolinones

Supporting Information). The results listed in Scheme 4 reveal that this one-pot synthetic method of pyrrolo[4,3,2-de]quinolinones 6 has a broad scope with respect to substrate 1. Variation of substituents R¹, R², and X in 1 is flexible and compatible. When R² was aryl or heteroaryl, 1 readily afforded the corresponding products 6a-p in excellent yields except 1j (X = H, R^1 = Bn, R^2 = 2-thienyl) only delivered its product 6q in a 21% yield. Other substrates 1 bearing representative substituents R¹ at the nitrogen atom or different substituents X on the indole framework all smoothly afforded their corresponding products 6r-z and 6A-F in moderate to excellent yields. Functional group ester substituted 1c and 1k ($R^1 = Me$, $R^2 = CO_2Et$, X = 6-Cl) were also effective, giving their corresponding products 6G and 6H in high yields. A gram-scale synthesis of 6a was also illustrated in a 92% yield with 1a (5.0 mmol) used (Scheme 4). Thus, this one-pot synthetic method constitutes a simple and efficient route to pyrrolo[4,3,2-de]quinolinones 6 with highly flexible substituents.

To further demonstrate the synthetic utility of this methodology, we developed a concise and highly efficient total synthesis of alkaloid ammosamide B (Scheme 5). Starting from compound

Scheme 5. Concise Total Synthesis of Ammosamide B

$$\begin{array}{c} \text{CO}_2\text{Et} \\ \text{N} \\ \text{N} \\ \text{O} \\ \text{OC} \\ \text{IN} \\ \text{OC} \\ \text{OC} \\ \text{IN} \\ \text{OC} \\$$

6H, ammosamide B can be readily obtained in three steps including nitration, reduction, and ammonolysis with an overall yield of 46.7%. According to Fenical's reports, the ammosamide B can be conveniently converted to ammosamides A and C through one or two more steps. Thus, this methodology provides simple and practical access to ammosamides A–C.

In summary, a novel annulation reaction between 3acylmethylidene oxindoles and in situ generated Huisgen zwitterions is disclosed, which represents a new annulation mode of Huisgen zwitterions involving a rare N-N bond cleavage. More importantly, this reaction unveils an unprecedented one-step strategy to efficiently construct complex pyrrolo[4,3,2-de]quinoline skeleton from readily available materials. On the basis of this annulation reaction, a one-pot synthetic methodology has been successfully developed, providing simple and practical access to a wide variety of pyrrolo[4,3,2-de]quinolinones and natural alkaloid ammosamides A-C as well. This robust methodology is accordingly anticipated to significantly facilitate the structure-activity relationship studies of natural pyrroloquinoline alkaloids and their analogues. Relevant investigations are currently underway in our laboratory.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b00456.

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Experimental details, characterization data, X-ray crystal structures for 3a,c, 4b, 5, 6a, and 7, and NMR spectra for new compounds (PDF)

Crystallographic file of product 3a(CIF)

Crystallographic file of product 3c(CIF)

Crystallographic file of product 4b(CIF)

Crystallographic file of product 5(CIF)

Crystallographic file of product 6a(CIF)

Crystallographic file of product 7(CIF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) For selected reviews, see: (a) Moilinski, T. F. Chem. Rev. 1993, 93, 1825. (b) Ding, Q.; Chichak, K.; Lown, J. W. Curr. Med. Chem. 1999, 6, 1. (c) Urban, S.; Hickford, S. J. H.; Blunt, J. W.; Munro, M. H. G. Curr. Org. Chem. 2000, 4, 765. (d) Antunes, E. M.; Copp, B. R.; Davies-Coleman, M. T.; Samaai, T. Nat. Prod. Rep. 2005, 22, 62. (e) Harayama, Y.; Kita, Y. Curr. Org. Chem. 2005, 9, 1567. (f) Hu, J.-F.; Fan, H.; Xiong, J.; Wu, S.-B. Chem. Rev. 2011, 111, 5465.
- (2) (a) Perry, N. B.; Blunt, J. W.; McCombs, J. D.; Munro, M. H. G. *J. Org. Chem.* **1986**, *51*, 5476. (b) Perry, N. B.; Blunt, J. W.; Munro, M. H. G. *Tetrahedron* **1988**, *44*, 1727. (c) Blunt, J. W.; Munro, M. H. G.; Battershill, C. N.; Copp, B. R.; McCombs, J. D.; Perry, N. B.; Prinsep, M.; Thompson, A. M. *New J. Chem.* **1990**, *14*, 761.
- (3) (a) Nagata, H.; Ochiai, K.; Aotani, Y.; Ando, K.; Yoshida, M.; Takahashi, I.; Tamaoki, T. *J. Antibiot.* 1997, 50, 537. (b) Aotani, Y.; Nagata, H.; Yoshida, M. *J. Antibiot.* 1997, 50, 543. (c) Nagata, H.; Yano, H.; Sasaki, K.; Sato, S.; Nakanishi, S.; Takahashi, I.; Tamaoki, T. *Biosci., Biotechnol., Biochem.* 2002, 66, 501.
- (4) (a) Gaudencio, S. P.; MacMillan, J. B.; Jensen, P. R.; Fenical, W. *Planta Med.* **2008**, *74*, 1083. (b) Hughes, C. C.; MacMillan, J. B.; Gaudencio, S. P.; Jensen, P. R.; Fenical, W. *Angew. Chem., Int. Ed.* **2009**, *48*, 725. (c) Hughes, C. C.; Fenical, W. *J. Am. Chem. Soc.* **2010**, *132*, 2528. (d) Hughes, C. C.; MacMillan, J. B.; Gaudencio, S. P.; Fenical, W.; La Clair, J. J. *Angew. Chem., Int. Ed.* **2009**, *48*, 728.
- (5) Reddy, P. V. N.; Jensen, K. C.; Mesecar, A. D.; Fanwick, P. E.; Cushman, M. J. Med. Chem. **2012**, 55, 367.
- (6) For selected examples, see: (a) Kita, Y.; Tohma, H.; Inagaki, M.; Hatanaka, K.; Yakura, T. J. Am. Chem. Soc. 1992, 114, 2175. (b) Sadanandan, E. V.; Cava, M. P. Tetrahedron Lett. 1993, 34, 2405. (c) White, J. D.; Yager, K. M.; Yakura, T. J. Am. Chem. Soc. 1994, 116, 1831. (d) Liang Tao, X.; Cheng, J.-F.; Nishiyama, S.; Yamamura, S. Tetrahedron 1994, 50, 2017. (e) Izawa, T.; Nishiyama, S.; Yamamura, S. Tetrahedron 1994, 50, 13593. (f) Sadanandan, E. V.; Pillai, S. K.; Lakshmikantham, M. V.; Billimoria, A. D.; Culpepper, J. S.; Cava, M. P. J. Org. Chem. 1995, 60, 1800. (g) Iwao, M.; Motoi, O.; Fukuda, T.; Ishibashi, F. Tetrahedron 1998, 54, 8999. (h) Tatsuta, K.; Imamura, K.; Itoh, S.; Kasai, S. Tetrahedron Lett. 2004, 45, 2847. (i) Wu, Q.; Jiao, X.; Wang, L.; Xiao, Q.; Liu, X.; Xie, P. Tetrahedron Lett. 2010, 51, 4806. (j) Reddy, P. V. N.; Banerjee, B.; Cushman, M. Org. Lett. 2010, 12, 3112. (k) Oshiyama, T.; Satoh, T.; Okano, K.; Tokuyama, H. Tetrahedron 2012, 68, 9376. (1) Yang, S.-W.; Wang, C.-M.; Tang, K.-X.; Wang, J.-X.; Sun, L.-P. Eur. J. Org. Chem. 2016, 2016, 1050.
- (7) For typical strategies to construct the pyrroloquinoline core, see: (a) Venemalm, L.; Esteves, C.; Alvarez, M.; Joule, J. A. *Tetrahedron Lett.*

1993, 34, 5495. (b) Balczewski, P.; Joule, J. A.; Estevez, C.; Alvarez, M. J. Org. Chem. 1994, 59, 4571. (c) Peat, A. J.; Buchwald, S. L. J. Am. Chem. Soc. 1996, 118, 1028. (d) Inoue, K.; Ishikawa, Y.; Nishiyama, S. Org. Lett. 2010, 12, 436. (e) Takayama, Y.; Yamada, T.; Tatekabe, S.; Nagasawa, K. Chem. Commun. 2013, 49, 6519. (f) Liu, Y.; Zhang, J.-L.; Song, R.-J.; Qian, P.-C.; Li, J.-H. Angew. Chem., Int. Ed. 2014, 53, 9017. (g) Shan, D.; Gao, Y.; Jia, Y. Angew. Chem., Int. Ed. 2013, 52, 4902. (h) Gao, Y.; Shan, D.; Jia, Y. Tetrahedron 2014, 70, 5136.

- (8) Pan, E.; Oswald, N. W.; Legako, A. G.; Life, J. M.; Posner, B. A.; MacMillan, J. B. Chem. Sci. 2013, 4, 482.
- (9) (a) Cookson, R. C.; Locke, J. M. J. Chem. Soc. 1963, 6039.(b) Huisgen, R.; Blaschke, H.; Brunn, E. Tetrahedron Lett. 1966, 7, 405.
- (c) Brunn, E.; Huisgen, R. Angew. Chem., Int. Ed. Engl. 1969, 8, 513.(d) Mitsunobu, O. Synthesis 1981, 1981, 1.
- (10) (a) Nair, V.; Menon, R. S.; Sreekanth, A. R.; Abhilash, N.; Biju, A. T. Acc. Chem. Res. **2006**, 39, 520. (b) Nair, V.; Biju, A. T.; Mathew, S. C.; Babu, B. P. Chem. Asian J. **2008**, 3, 810. (c) Xu, S.; He, Z. RSC Adv. **2013**, 3, 16885.
- (11) For selected reports, see: (a) Otte, R. D.; Sakata, T.; Guzei, I. A.; Lee, D. Org. Lett. 2005, 7, 495. (b) Nair, V.; Biju, A. T.; Vinod, A. U.; Suresh, E. Org. Lett. 2005, 7, 5139. (c) Nair, V.; Mathew, S. C.; Biju, A. T.; Suresh, E. Angew. Chem., Int. Ed. 2007, 46, 2070. (d) Cui, S.-L.; Wang, J.; Wang, Y.-G. Org. Lett. 2008, 10, 13. (e) Lian, Z.; Guan, X.-Y.; Shi, M. Tetrahedron 2011, 67, 2018. (f) Sankar, M. G.; Garcia-Castro, M.; Wang, Y.; Kumar, K. Asian J. Org. Chem. 2013, 2, 646. (g) Nair, V.; Biju, A. T.; Mohanan, K.; Suresh, E. Org. Lett. 2006, 8, 2213. (h) Chakravarty, M.; Kumar, N. N. B; Sajna, K. V.; Swamy, K. C. K. Eur. J. Org. Chem. 2008, 2008, 4500. (i) Hong, D.; Zhu, Y.; Lin, X.; Wang, Y. Tetrahedron 2011, 67, 650.
- (12) Yang, C.; Li, J.; Zhou, R.; Chen, X.; Gao, Y.; He, Z. Org. Biomol. Chem. 2015, 13, 4869.
- (13) According to our previous report, product **5** is thought to form through a sequence of double-nucleophilic additions followed by elimination of Ph₃PO. See ref 12.
- (14) In the phosphine-mediated (aza)cyclopropanations of electron-deficient alkenes or imines with aldehydes or ketones, similar mechanisms are also thought to take effect. See: (a) Liu, X.-G.; Wei, Y.; Shi, M. Eur. J. Org. Chem. 2010, 2010, 1977. (b) Liu, X.-G.; Wei, Y.; Shi, M. Tetrahedron 2010, 66, 304. (c) Xu, S.; Zhou, L.; Ma, R.; Song, H.; He, Z. Org. Lett. 2010, 12, 544 and references cited therein.
- (15) Some diaziridines with functional groups incorporated in the substituents are very unstable. Makhova, N. N.; Petukhova, V.; Yu; Kuznetsov, V. V. *ARKIVOC* **2008**, 128 and references cited therein.
- (16) A similar electrophilic aromatic substitution with an in situ generated nitrenium ion is proposed in the synthesis of 2-oxindoles from *N*-chloro-*N*-methoxyamides. Kikugawa, Y.; Kawase, M. *J. Am. Chem. Soc.* **1984**, *106*, 5728.
- (17) The in situ generation of Huisgen zwitterions is confirmed by the NMR monitoring experiments. For details, see the Supporting Information.
- (18) Counting from the commercially available 6-chloroisatin, ammosamide B was synthesized in seven simple steps with an overall yield of 12.9%. A schematic synthetic route is available in the Supporting Information. This route currently represents the most efficient synthesis of ammosamide B. For the total synthesis of ammosamide B, see refs 4c, 6i,j,l, and 7e.